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Network

Nanocomposite for the  
Selective Removal of  
Pb(II) and Hg(II) from  
Synthetic Wastewater.  
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Electrochemical  
Biosensor of Nanocube-  
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periodic table will be  
given. Bring a  
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concepts from General  
Chemistry 1 & 2,  
Quantitative Analytical,  
and Organic I/II e.g.  
prefixes nm, cm, Å,  
etc. Chapter 1  
Fundamental particles  
of an atom Atomic  
number, mass number,  
notation Relative  
atomic mass and  
isotopes, weighted

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average mass

spectroscopy and

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Relationships ...

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If I recall correctly, with the inorganic and p-chem ACS finals there is a question bank that your professor chooses from (in part, some

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questions are on all exams), so it will be tailored to material you learned in class. So for example my class had questions on organo-metallic catalysis and bio-inorganic that were particularly challenging.

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qualifying exams we use at Clark are based on the typical advanced inorganic chemistry undergraduate course as taught in most American Universities. Textbooks such as the following adequately cover the material tested by the exams. Cotton, Wilkinson and Gaus, "Basic Inorganic Chemistry"

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**CHEMISTRY AND  
BIOCHEMISTRY**

Inorganic chemistry is concerned with the properties and behavior of inorganic compounds, which include metals, minerals, and organometallic compounds. While organic chemistry is defined as the study of carbon-containing compounds and inorganic chemistry is the study of the

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remaining subset of compounds other than organic compounds, there is overlap between the two fields (such as organometallic compounds, which usually contain a metal or metalloid bonded directly to carbon).

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- American Chemical  
Society**

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quantum chemistry are  
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Elements. Gold (18512)

Oxygen (17767)

Carbon (14828)

Hydrogen (11648)

Platinum (10435)

Palladium (9514)

Nitrogen (8922) Silicon

(8777) Iron (6795)

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